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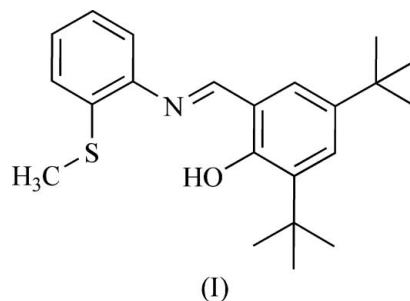
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Key indicators

Single-crystal X-ray study
 $T = 173$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.047
 wR factor = 0.147
Data-to-parameter ratio = 20.0For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.3,5-Di-*tert*-butyl-*N*-[2-(methylsulfanyl)phenyl]-
salicylaldimineThe molecule of the title compound, $\text{C}_{22}\text{H}_{29}\text{NOS}$, is non-planar, with a dihedral angle of 32.06 (9) $^\circ$ between the two aromatic rings.Received 6 July 2006
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Comment

Schiff base ligands are routinely used in coordination chemistry. The choice of ligand for a metal complex is crucial in determining its chemical and physical properties. Our group is interested in the synthesis and preparation of novel sulfur-containing Schiff base ligands for use in coordination chemistry (Halbach & Hamaker 2006; Hamaker & Oberts, 2006). As part of our ongoing studies, we report the synthesis, properties and crystal structure of the title compound, (I).

Unlike the 3-methoxy derivative (Hamaker & Corgliano, 2006), the molecule of (I) is non-planar (Fig. 1), with a dihedral angle between the two aromatic rings of 32.06 (9) $^\circ$. The bond lengths and angles (Table 1) are similar to those of related molecules (Hamaker & Corgliano, 2006; Hamaker *et al.*, 2006). Similar to the 3-methoxy derivative, there is an intramolecular $\text{O1}-\text{H1}\cdots\text{N1}$ hydrogen bond (Table 2).

Experimental

A solution of 2-(methylsulfanyl)aniline (1.84 g, 13.2 mmol) in 30 ml ethanol was placed into a 150 ml flask equipped with a reflux condenser. To this solution, 3,5-di-*tert*-butyl-2-hydroxybenzaldehyde (3.01 g, 12.9 mmol) was added. The mixture was refluxed for 1 h and then cooled to room temperature, during which time a light-yellow precipitate formed. The mixture was then placed in a freezer at 233 K for 3 d. The solid was collected by suction filtration, washed with ice-cold ethanol and air-dried, resulting in shiny yellow microcrystals (yield: 3.648 g, 86%). Single crystals of (I) were grown by slow evaporation of a saturated ethanol solution. IR (Nujol, ν , cm^{-1}): 3577 (OH); 1612 (*s*, $\text{N}=\text{C}$). Analysis calculated (found) (%) for $\text{C}_{22}\text{H}_{29}\text{NOS}$: C 74.32 (74.37), H 8.22 (8.26), N 3.94 (4.10).

Crystal data

C₂₂H₂₉NOS
M_r = 355.52
 Monoclinic, *P*2₁/*c*
a = 17.876 (3) Å
b = 8.990 (2) Å
c = 12.6963 (12) Å
 β = 98.161 (8)°
V = 2019.7 (6) Å³

Z = 4
D_x = 1.169 Mg m⁻³
 Mo *K*α radiation
 μ = 0.17 mm⁻¹
T = 173 (2) K
 Prism, yellow
 0.48 × 0.37 × 0.37 mm

Data collection

Enraf–Nonius CAD-4
 diffractometer
 $\omega/2\theta$ scans
 Absorption correction: none
 4796 measured reflections
 4589 independent reflections

3291 reflections with *I* > 2σ(*I*)
*R*_{int} = 0.041
 θ_{\max} = 27.5°
 3 standard reflections
 frequency: 120 min
 intensity decay: 4%

Refinement

Refinement on *F*²
R[*F*² > 2σ(*F*²)] = 0.047
wR(*F*²) = 0.147
S = 1.04
 4589 reflections
 230 parameters
 H atoms treated by a mixture of
 independent and constrained
 refinement

$w = 1/[\sigma^2(F_o^2) + (0.0762P)^2 + 1.1485P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.41 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.32 \text{ e } \text{Å}^{-3}$

Table 1

Selected geometric parameters (Å, °).

C1–S1	1.792 (2)	C7–N1	1.413 (2)
C2–S1	1.760 (2)	C10–N1	1.281 (2)
N1–C10–C11	121.88 (18)	C2–S1–C1	102.46 (11)
C10–N1–C7	122.34 (17)		

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
O1–H1···N1	0.80 (4)	1.90 (4)	2.571 (2)	142 (3)

The H atom of the hydroxy group was found in a difference Fourier map and refined isotropically. The other H atoms were treated as riding atoms, with C–H distances of 0.95 (aromatic) or 0.98 Å (aliphatic), and with *U*_{iso}(H) = 1.2*U*_{eq}(C) or 1.5*U*_{eq}(methyl C). The displacement ellipsoids for the methyl carbons of the C30–C33

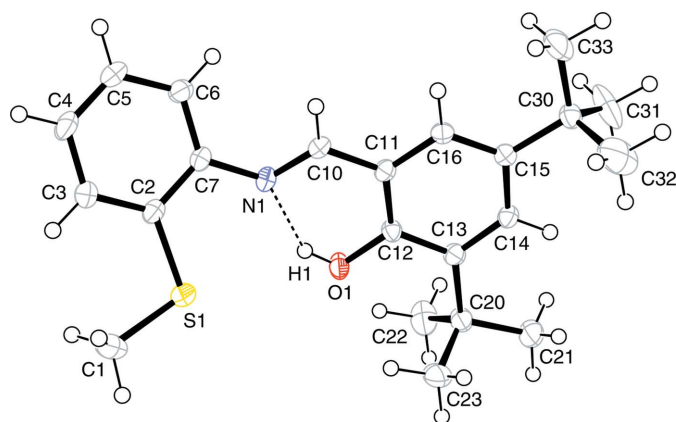


Figure 1

The molecular structure of (I) with 50% probability displacement ellipsoids. The dashed line indicates the hydrogen bond.

tert-butyl group are elongated, suggesting possible disorder. Attempts to model the *tert*-butyl group as disordered over two sites did not give satisfactory results. The elongated displacement ellipsoids reflect partial free rotation of the *tert*-butyl group.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *DIRDIF99* (Beurskens *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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